

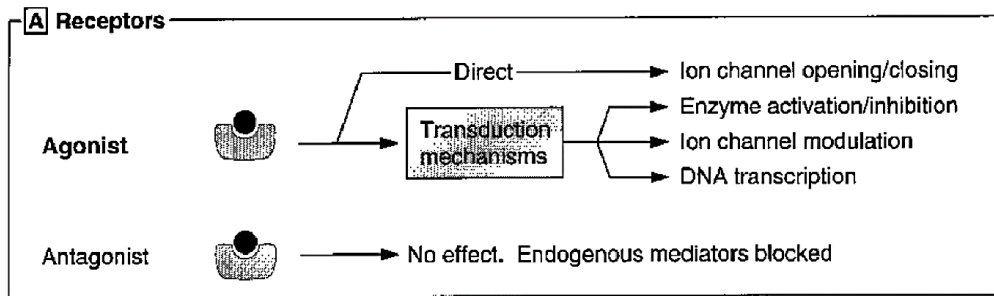
## » Drug Targets 1: Protein Targets for Drug Action

- Protein targets for drug action: **RICE**

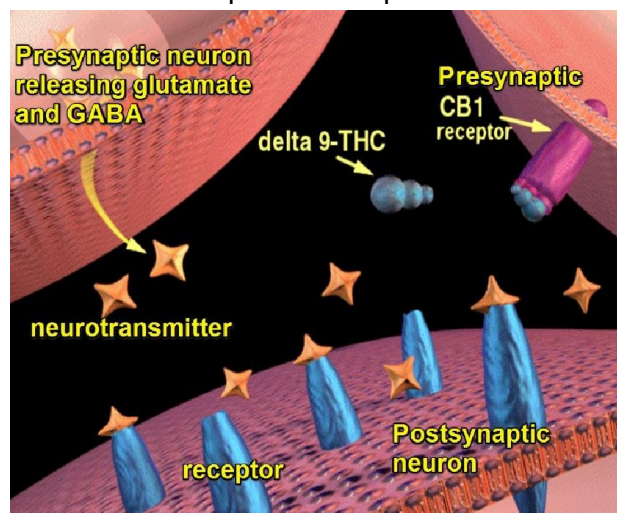
1. Receptors
2. Ion channels
3. Carriers
4. Enzymes

### # Receptors

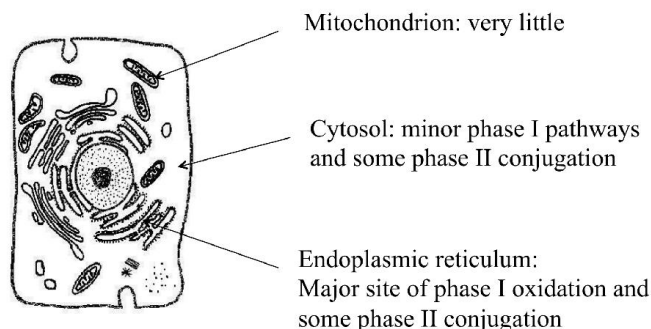
- Proteins found in the cell membrane
- Responsible for receiving chemical information from hormones, transmitters, cytokines and growth factors.



- Cannabis: **CB1 receptor** (cannabinoid receptor type 1)
  - Cannabis is the most widely used illicit drug in the world.
  - The main psychoactive constituent (cannabinoid) is  $\Delta^9$ -THC.
    - $\Delta^9$ -THC binds to CB1 receptors in the brain, producing classical cannabinoid effects (euphoria, anxiety, memory impairment and appetite stimulation).
  - Mechanism of action:
    - This is a synapse containing the endogenous cannabinoid system. The CNS contains CB1 receptors and endogenous mediators (endocannabinoids) such as anandamide and 2-AG. These endocannabinoids are physiological ligands for the cannabinoid receptors.
    - Glutamate acts on glutamate receptor proteins in the postsynaptic membrane, and is responsible for 'activating' a person.
      - × **Glutamate is the main excitatory neurotransmitter.** It depolarises neurons by increasing membrane  $\text{Na}^+$  conductance.
    - The presynaptic CB1 receptors are autoreceptors – i.e. this system modulates the release of other neurotransmitters.
      - ×  $\Delta^9$ -THC mimics the endogenous mediator anandamide by binding to CB1 receptors and inhibiting the release of glutamate (presynaptic inhibition).
    - This reduces 'activation' of a person and produces a calming effect.



- Sites of biotransformation in the cell:

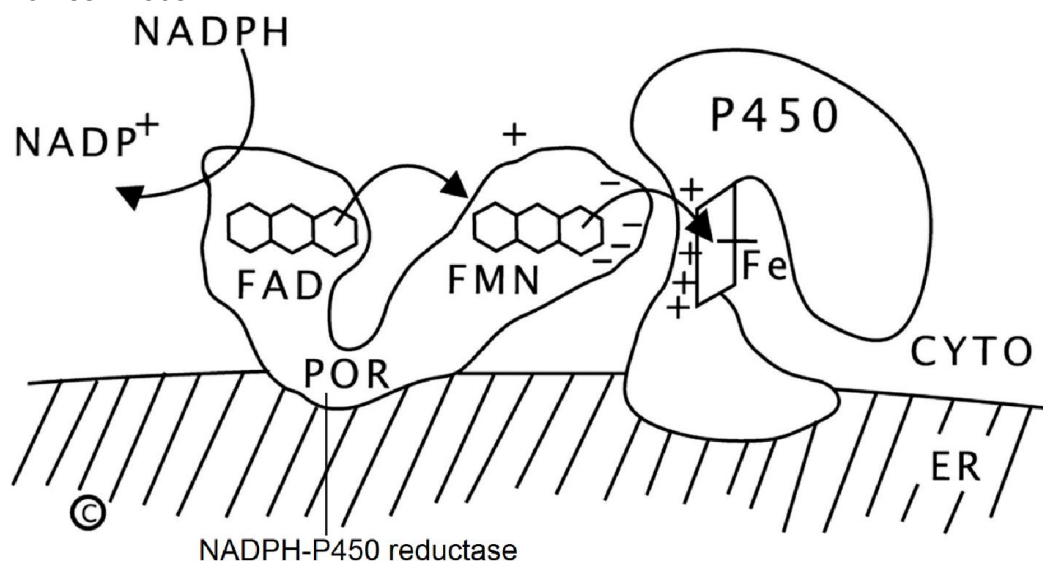


- Phase I reactions:

- Chemical conversion of a lipophilic chemical into a more polar analogue.
- Inclusion of a new functional group (usually by oxidation, reduction or hydrolysis), which is subject to phase II metabolism.
  - Example: Oxidation of the O-ethyl group in phenacetin produces a phenol, with ~10 fold decrease in lipophilicity.

## # Cytochromes P450

- **P450** (aka **CYP**) is the major class of phase I biotransformation enzymes.
- Unlike most enzymes, **P450s are multifunctional enzymes that act on diverse substrates**, including the xenobiotics and endobiotics to which we are exposed.
  - **P450s have low substrate specificity** – can accommodate a wide range of substrates.
  - However, there are still some drugs and chemicals that are oxidised by a single P450.
- Factors that affect P450 activities can markedly affect drug elimination.
- Components of the Phase I P450 System: Lipophilic xenobiotics and endobiotics are oxidised by cytochromes P450s.

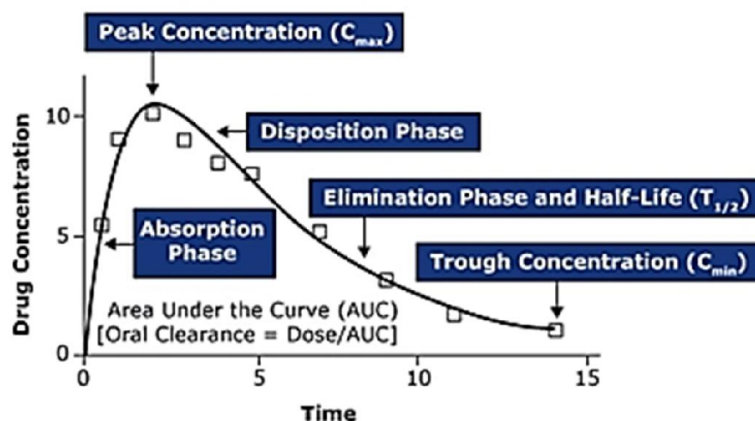


- **P450 enzyme**
  - Site of oxygen activation (activated oxygen inserted into substrate)
- **NADPH**
  - Source of electrons that drives activation of oxygen.
- The cofactor **NADPH-P450 reductase**
  - Transfers electron from NADPH to P450 enzyme.
- **Membrane phospholipid**
  - To reproduce the environment within the endoplasmic reticulum.
- Characteristic P450 Reactions:
  - Drugs often form several metabolites because they contain more than one substituent that can be oxidised.

## # Pharmacokinetic Parameters

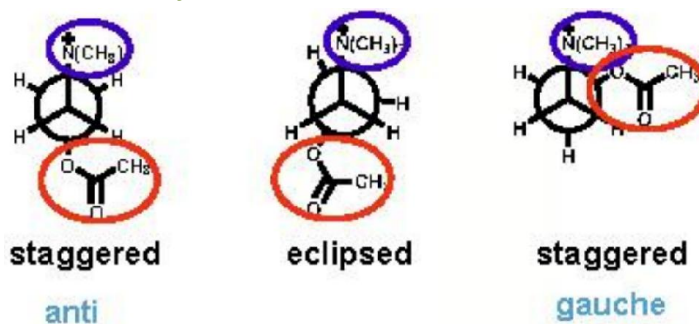
- After absorption, a drug builds up effective serum concentrations, but is also converted to forms that are more readily eliminated.
- The characteristics of these processes are described by a series of pharmacokinetic parameters.

### Pharmacokinetic curve of a drug in the body



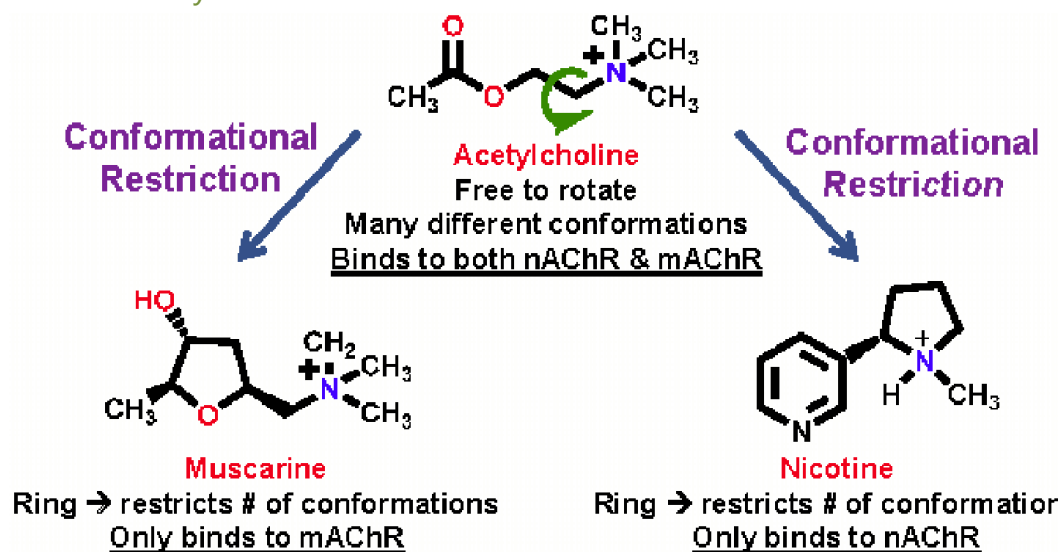
Pharmacokinetic Parameter	Definition
$C_{\max}$ ( $\mu\text{g/mL}$ )	<ul style="list-style-type: none"> <li>• <b>Peak plasma concentration (<math>C_p</math>)</b> <ul style="list-style-type: none"> <li>– Usually following an oral dose</li> </ul> </li> <li>• Reflects drug disposition in terms of bioavailability and distribution</li> </ul>
$T_{\max}$ (h)	<ul style="list-style-type: none"> <li>• <b>Time taken to reach <math>C_{\max}</math></b></li> </ul>
$C_{\min}$ ( $\mu\text{g/mL}$ )	<ul style="list-style-type: none"> <li>• <b>Trough <math>C_p</math></b> at the end of the dosing interval (before the next dose is given)</li> </ul>
<b>Clearance</b> (L/h)	<ul style="list-style-type: none"> <li>• <b>Volume of plasma cleared of the drug per unit time</b></li> <li>• Measure of efficiency of drug elimination <ul style="list-style-type: none"> <li>– Irreversible elimination from circulation</li> <li>– One-way elimination (sweat, urine, etc.) and/or metabolic conversion</li> </ul> </li> <li>• At a given IV dose rate, clearance is the sole parameter determining steady state <math>[\text{drug}]_{\text{plasma}}</math></li> </ul>
<b>Volume of distribution</b> (L or L/kg)	<ul style="list-style-type: none"> <li>• <math>V_d = \frac{\text{dose (g)}}{\text{initial plasma concentration, } C_0 \text{ (g/L)}}</math></li> <li>• Relates amount of drug in body to <math>[\text{drug}]_{\text{plasma}}</math></li> <li>• Measure of extent of distribution <ul style="list-style-type: none"> <li>– The higher the <math>V_d</math>, the greater the proportion of drug that is distributed outside the blood stream</li> </ul> </li> <li>• <math>V_d</math> is not a real volume; it is the apparent volume into which the drug appears to distribute to achieve the measured <math>C_0</math></li> <li>• Used to calculate loading dose (initial dose required to achieve a given <math>C_p</math>) <ul style="list-style-type: none"> <li>– Back-calculated from <math>[\text{drug}]_{\text{plasma}}</math> following given dose</li> </ul> </li> </ul>
<b>Half-life</b> (h)	<ul style="list-style-type: none"> <li>• <b>Time to halve drug amount in body</b> <ul style="list-style-type: none"> <li>– Duration of action following single dose</li> </ul> </li> <li>• Determined by clearance (CL) and <math>V_d</math> <ul style="list-style-type: none"> <li>– <math>t_{1/2} = 0.693 \times \frac{V_d}{CL}</math></li> </ul> </li> <li>• Clearance follows first order kinetics (mostly exponential) <ul style="list-style-type: none"> <li>– <math>C_t = C_0 \times e^{-kt}</math></li> <li>– Doubling the dose only increases duration by one half-life</li> </ul> </li> </ul>
$AUC_{0-x}$ ( $\mu\text{g}\cdot\text{h/mL}$ )	<ul style="list-style-type: none"> <li>• <b>Area under concentration-time curve</b></li> </ul>

## # Flexibility of ACh



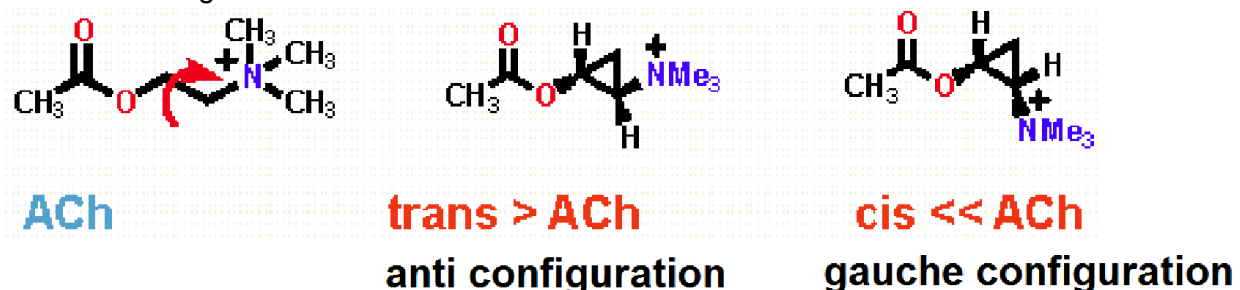
- Staggered conformation much more energetically stable than eclipsed.
- Staggered gauche conformation more stable than staggered anti because of stabilising intramolecular interaction between negative dipole of ester group and positive quaternary nitrogen.
  - Pulls energy of gauche conformer below energy of anti conformer.

## # Selectivity of Muscarine and Nicotine



## # Muscarinic Agonists

- Rotational flexibility about C-C bond can be restricted by installing cyclopropane rings.
- Trans and cis isomers interconvert the position of the quaternary amine.
  - Trans (but not cis) isomer is a much more potent agonist than ACh at the mAChR.
    - Depends on whether  $N^+Me_3$  and  $OAc$  groups are closer to anti than gauche configurations.



## » S-A Relationships for Cholinergic Drugs 2

### # Acetylcholinesterase Inhibitors

- Compounds which inhibit the activity of AChE cause a build up in the concentration of ACh in the synaptic cleft, which can result in an overstimulation of the cholinergic receptors in the PNS and CNS.
- Acetylcholinesterase inhibitors have toxic and therapeutic uses.