

Adverse drug reactions

- Noxious, unintended or undesired response to drug, caused by normal use of normal doses
- Direct causal relationship between the medicine and the occurrence

ADR	Characteristics
Type A (80%)	<ul style="list-style-type: none"> - Predictable and dose-dependent, usually related to pharmacological actions of drug - Immune system is <i>not</i> involved <ul style="list-style-type: none"> o E.g. Toxicity, intolerance, secondary effects, drug interactions, renal failure - Treatment – reduce dose or withhold drug
Type B (15%)	<ul style="list-style-type: none"> - Unpredictable, not dose-dependent - Unrelated to pharmacological actions of the drug, e.g. penicillin allergy - Treatment – withhold drug and avoid rechallenge

Type B ADR	Antibodies involved	Examples	Timeline
1 – immediate	IgE	Dyspnoea, urticarial, swelling, anaphylaxis	Hours
2 – cytotoxic	IgM, IgG	Haemolytic anaemia, thrombocytopenia	5-14 days
3 – immune complex	IgM, IgG	Small vessels – nephritis, vasculitis, iritis	7-8 days
4 – cell-mediated	None; involves T-cells	Skin rash, eczema, maculopapular, bullous	Weeks

Type 4: delayed-type reactions (most common)

- T_H1 produces TNF- α and IFN- γ , activates macrophages
- T_H2 produces interleukins, activates B-cells
- Drugs include – abacavir, allopurinol, azathioprine, NSAIDs, terbinafine, quinolones, sulfonamides, beta-lactam antibiotics

Anaphylaxis

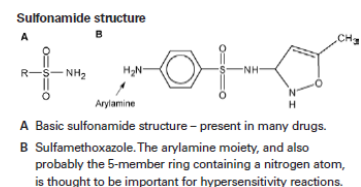
- Combination of immediate-type reactions – usually occurs within 15 minutes to hours
 - o Respiratory symptoms – cough, wheeze, hoarseness, cyanosis, upper airway swelling
 - o Cardiovascular – tachycardia, severe hypotension, loss of consciousness, palpitations
 - o Dermatological – urticarial, erythema, angioedema
 - o Nausea, vomiting, abdominal cramps, anxiety, distress
- Treatment – adrenaline IM injection into thigh (blue for the sky, orange for the thigh) + hospital
 - o Hold autoinjector in place for 10 seconds, repeat dose every 5 minutes until improvement
 - Lay patient flat, ensure open airways
 - o Antihistamines and hydrocortisone are *not* first-line (but may be given later on)
- Causes of anaphylaxis – medicines > insect venom > foods

Haptens

- Antigen too small (insufficient molecular weight) to generate immune response on its own
 - o An antigen must be (or attach to) a macromolecule, e.g. large protein, polysaccharide
- Metabolism may produce reactive metabolites haptenate endogenous proteins (increases MW)

Antibiotic allergies

- May affect any organ, but most commonly the skin
- Higher rates in HIV and infectious mononucleosis patients
- Patients with history of penicillin allergy who require cephalosporins
 - o Possible cross-sensitivity as both are beta-lactam antibiotics
 - o Treatment depends on if previous reaction was IgE-mediated, and severity of reaction
- Sulfur allergy – no evidence of cross-reactivity between arylamine and non-arylamine sulfonamides
 - o Presence of arylamine ring in sulfonamide structure conveys immunogenicity
 - o Arylamine sulfonamides – allergic cross-sensitivity within this groups is possible
 - Many antibiotics, sulfasalazine, sulfamethoxazole, amprenavir, fosampranavir
 - o Non-arylamine sulfonamides – no allergic cross-reactivity
 - Acetazolamide, sulfonyleureas, loop diuretics, thiazide diuretics, celecoxib
- Allergy to sulfonamides – do *not* need to avoid sulfites/sulfates/sulfur (since low MW)

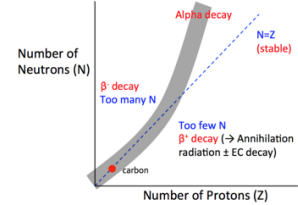


Term	Definition
Nuclide	Atomic species identified by exact nuclear composition, including mass #A and atomic #Z
Radionuclide	Unstable or radioactive nuclide
Isotopes	Same atomic #Z but different neutron number = different mass #A
Isomers	Same number of protons and neutrons, but different energy states

$$\begin{matrix} \text{Mass number} \\ = \# \text{ of protons} + \# \text{ of neutrons} \\ A \\ \times \\ \text{Atomic number} \\ = \# \text{ of protons} \\ Z \end{matrix}$$

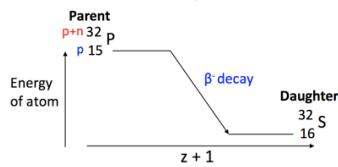
Radioactive decay

- 81 elements with ≥ 1 isotope = 274 nuclides (most of which are unstable)
- Heavier nuclides tend to have more neutrons than protons
- Unstable nuclides – mostly produced in nuclear reactor or accelerator
- Cyclotron – particle accelerator
 - o Used to bombard neutron-deficient target with protons
 - o Produces *positron*-emitting radioisotopes

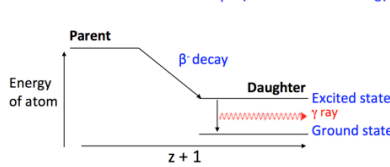


Nuclear instability	Decay mechanism	Mode of decay (i.e. emission)
Excess neutrons	$n \rightarrow p^+ + e^- + \nu$	Electron (β^-) + neutrino (ν)
Excess protons (neutron deficiency)	$p^+ \rightarrow n + e^+$	Positron (β^+) ...or...
	$p^+ + e^- \rightarrow n + \nu$	Electron capture
Excess neutrons + protons	${}^A_Z X \rightarrow {}^{A-4}_{Z-2} X + {}^4_2 \text{He}$	Alpha (${}^4_2 \text{He}$)

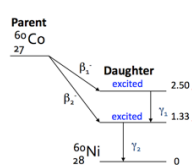
Excess neutrons $n \rightarrow p^+ + e^- + \nu$



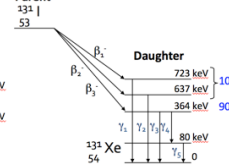
Gamma radiation emission of γ rays to lose excess energy



Decay of ${}^{60}\text{Co}$



Decay of ${}^{131}\text{I}$

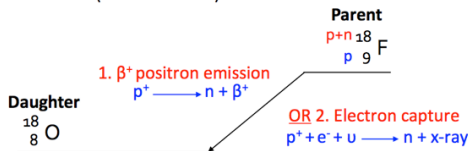


Excited states

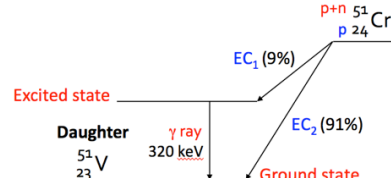
- May be multiple excited states
 - o ${}^{60}_{27}\text{Co}$ can decay to 2.50 or 1.33 MeV to emit two different energies of γ rays; ${}^{131}_{53}\text{I}$ has four
- More convenient for imaging if there is only one level of gamma ray energy to detect

Nuclides with too FEW neutrons

F \rightarrow O (but same mass)



Electron capture Decay of ${}^{51}\text{Cr}$

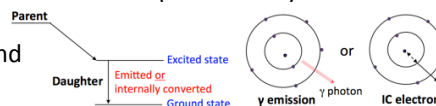


Annihilation (gamma) radiation

- After a positron leaves the nucleus, it loses kinetic energy and is annihilated by combining with an electron to form 2 photons (2 gamma rays)
 - o $\beta^+ + \beta^- \rightarrow 2\gamma$
 - o I.e. mass of the particles is converted to electromagnetic radiation ($E=mc^2$)
- Both photons have energy of 0.51 MeV, but travel in opposite directions (\therefore net momentum = 0)
 - o Since photons are emitted 180° to each other, they can be used in diagnostic scans

Electron capture (K-capture)

- Proton-rich nuclide absorbs/captures electron from an atomic orbital (usually K-shell)
 - o $p^+ + e^- \rightarrow n + \text{neutrino emission}$
- Excited state nuclide will emit gamma rays to return to ground state – two possible ways
 1. Emit energy directly
 2. Internal conversion – energy is transferred to a bound electron, which is then ejected from the atom



Chemotherapy agents

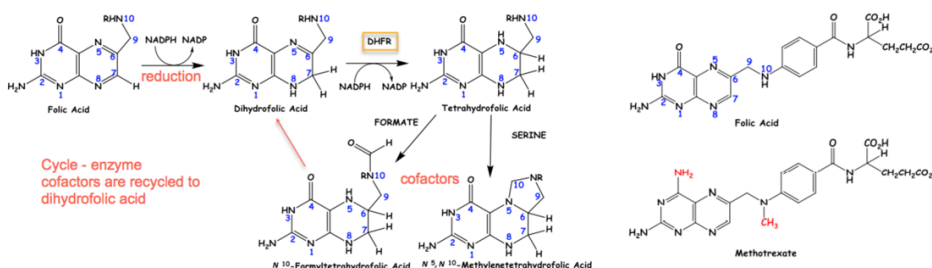
Cell cycle	G₀ → interphase (G₁ → S → G₂) → M
G ₀	Quiescence – cells can exit cell cycle in G ₁ to enter G ₀
G ₁ , G ₂	Gap phases – rest, cellular growth and synthesis
S phase	Synthesis – DNA replication <ul style="list-style-type: none"> - DNA helicase – unwinds and unzips helix - DNA polymerase – inserts new nucleotides - DNA ligase – links sugar-phosphate groups between new nucleotides
M phase	Mitosis – cell division

Anti-metabolites (target S-phase)

- Cancer cells – grow rapidly, require more nucleotides than normal cells for DNA synthesis
 - o Purine bases (AG) and pyrimidine bases (CTU)
 - Obtained *de novo* and via salvage pathways – all cells use both
 - Cancer cells favour *de novo* synthesis
- Antimetabolites prevent biosynthesis or utilisation of normal cellular metabolites
 1. May be enzyme inhibitor or fraudulent building block
 2. Incorporated into RNA/DNA to inhibit *de novo* synthesis of nucleotides
 3. Affects replication, transcription and translation
- E.g. pyrimidine analogues (5-fluorouracil), purine analogues (6-mercaptopurine), methotrexate
 - o Sufficiently similar to essential metabolite – mistaken for, but cannot take its place
 - o Usually result from ≥1 bioisosteric changes (similar size but different properties)
 - E.g. H → F, CH₂/O → S, OH → NH₂

Methotrexate (MTX)

- Uses – leukaemia
- Resistance due to up-regulation of DHFR gene → insufficient MTX to inhibit all of DHFR
- MOA – competitive dihydrofolate reductase (DHFR) inhibitor
 - o Folic acid – essential metabolite for biosynthesis of DNA
 - o DHFR is essential in regeneration of folic acid cofactors required for DNA synthesis
 - o DHFR inhibition → absence of cofactor → inhibition of purine and pyrimidine synthesis
- Folic acid → dihydrofolic acid + DHFR → tetrahydrofolic acid → cofactors
- Cofactors – 1 carbon donors (formyl CHO or methyl CH₃) necessary to produce DNA bases
 - o N¹⁰-formyl-FH₄ – for purine synthesis
 - o N⁵N¹⁰-methyl-FH₄ – for pyrimidine synthesis
 - Thymidylate synthetase normally uses dUMP + cofactor to produce dTMP



5-fluorouracil (5FU)

- Uses – breast and stomach cancers; causes bone marrow toxicity
- Resistance due to down-regulation of cellular enzymes needed to add deoxyribosephosphate unit
- 5FU only inhibits thymidylate synthetase – blocks conversion of uracil (dUMP) to thymine (dTMP)
 - o Differs from dUMP by replacing H with F (more electronegative) to become dFUMP
- Thymidylate synthetase binds N⁵N¹⁰-methyl-FH₄ cofactor + dFUMP
 - o Produces irreversibly-bound enzyme-substrate complex with the fluorine
 - o Prevents pyrimidine and DNA synthesis

