## THE STRUCTURE AND FUNCTION OF LARGE BIOLOGICAL MOLECULES

## Molecules of Life:

- Lipids
- Carbohydrates \*
  - → Monosaccharide monomers

Proteins **←** → Amino acid monomers

Nucleic acids

→ Nucleotide monomers

**MACROMOLECULES** 

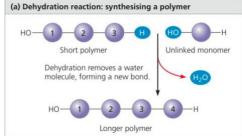
## Macromolecules:

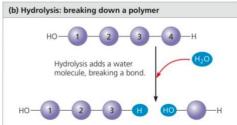
Polymer: a long-chain molecule consisting of repeating subunits called monomers

• Macromolecule: a large polymer made of smaller molecules joined together usually by dehydration reaction

Synthesis:

- → Dehydration reaction: a reaction in which two molecules are **covalently** bonded together, with the loss of a water molecule (enzyme catalysed)
- → Each molecule contributes part of the water molecule to be expelled
- Decomposition:
  - → Hydrolysis: a decomposition reaction where a water molecule is added to break a covalent bond (enzyme catalysed)
    - ⇒ E.g. digestion of food





## Carbohydrates:

- Carbohydrate: naturally occurring compounds of carbon, hydrogen and oxygen
  - → Monosaccharides
    - ⇒ Simplest sugars (usually some multiple of CH<sub>2</sub>O)
    - ⇒ Monomers from which more complex carbohydrates are built
    - $\Rightarrow$  E.g. glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>)
  - $\rightarrow$  Disaccharides
    - ⇒ 2 monosaccharides joined by a glycosidic bond
    - ⇒ E.g. maltose
  - → **Polysaccharides** (carbohydrate macromolecules)
    - ⇒ Many monosaccharides joined by glycosidic bonds
    - $\Rightarrow$  E.g. starch
- Glycosidic bond: a covalent bond formed between two monosaccharides by a dehydration reaction

